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Calibration of the Reacting Vessel of a Downdraft Gasifier for Temperature Optimization: Initial Validation

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Abstract: To achieve a tar free syngas production, the design and process variables of a gasifier must be considered and calibrated for optimum performance. This has been a major challenge in most gasification operation establishing the operating temperature with sensors considering the operational environment and duration. The research work presents a simple but valid process for the initial determination and calibration of temperature which can be done for any type of gasifier. This experimental calibration was done to establish the operating conditions of a laboratory scale fixed bed, downdraft gasifier for the conversion of wood waste to gaseous fuel using air as a medium by monitoring product parameters. The temperature, bed material, air flow ratio, nature and composition of the pellets and the product of the operation were considered for various trials. High temperature was achieved by increasing the refractory material used at the reduction zone of the bed and introducing char as bed material thus ensuring that the pyrolytic char is reduced with little or no tar in the product gas.

Keywords: Calibration, Downdraft gasifier, Temperature, Operating conditions, waste conversion.

1. INTRODUCTION

Gasification is a thermochemical conversion of biomass or other organic mass to a low heating value gas in an oxygen deficient environment. (Balas et al., 2007; Bhavana and sastry, 2011). This process generates renewable energy from biomass feedstock that can stand as an alternative to fossil fuel energy (Sutton et al). Gasification aims at transforming the largest amount of energy contained in the feed fuel (biomass wood waste) into the largest possible amount of energy in a gaseous form (Busa, 2013; Balas et al., 2007). This happens in a series of reaction that occur at an elevated temperature of about 750°C – 1000°C (Bhavanam Sastry 2011). The conversion of biomass using air as the reacting medium gives a mixture of gases called producer gas. Some of which are combustible [Syngas (CO, H₂ and CH₄)] and some are non – combustible (higher hydrocarbons, tar, N₂ and impurities) (Zhon et al., 2015; Bhavanam and Sastry, 2011; Balas et al., 2007). Air is the most commonly used gasification medium both for commercial and laboratory demonstration. These exit gases have an expected volumetric rate of about 50% N₂, 8% -20% H₂, 8% -20% CO, and 2% - 4% CH₄ with CO₂ and H₂O as the remaining volume (Akudo et al., 2012).

The design and operation of a gasifier require an understanding of the gasification process, its configuration, size, feedstock, and operating parameters influence on the performance of the plant. Thermo-chemical gasification of biomass is a well-known technology that can be classified depending on the gasifying agent: air, steam, steam–oxygen, air– steam, O_2 -enriched air, etc. Several types of gasifiers have been studied and developed; these gasifiers have different hydrodynamics (especially the way in which the solid fuel and the gasification agent are contacted), gasification agents (air, oxygen and/or steam) and operating conditions such as temperature and pressure.

The most important types are fixed-bed gasifier, (downdraft gasifier and updraft gasifier operated in counter-current, co-current or cross-current mode), fluidized bed gasifier and entrained flow gasifier. Downdraft gasifiers try to avoid the disadvantage of high tar contents by injecting the gasification agent not at the bottom but to a certain height above the bottom (Fjellerup et al., 2005). The main difference to updraft gasifier is that the gas flows co-currently downwards with the biomass. This leads to an order of the reaction zones from top to bottom with the result of low tar content in the fuel gas, namely, **drying, pyrolysis, oxidation**, and **reduction zone.** In any type of gasification operation, the

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reduction zone is the temperature controlling part of the gasifier type. The general performance and longevity of any gasifier is a function of the temperature of the reduction zone (Jia et al., 2013).

Tar is a major problem in a gasification process that can hinder the effective generation and use of the product gas (Balas et al., 2007; Bhavanam Sastry 2011; Akudo et al.,2012). To achieve a high conversion of the biomass and a low tar content insitu , the operating conditions must be properly selected, bed additives or catalyst can be used and a general gasifier modifications (Devi et., 2003). This can only be obtained by calibration and optimization. The rate determining step in a gasification process is the reaction between the char formed at the pyrolysis stage. The gaseous environment depends largely on variables such as pressure, temperature, particle size, mineral content and porosity. If these parameters are not managed well, more tar will be obtained (Asadullah,2014). Tar is an undesirable aromatic by-product having condensable hydrocarbon complex polyaromatic hydrocarbon and oxidized 1 - 5 aromatic ring. To produce a clean gas by the destruction of these tar component, temperature and other process parameters must be considered (Basu,2006; Koukouzas et al, 2008; Brandin et al., 2011) The parameters include gasifier type, pressure, bed material, equivalent ratio ER, etc.

Temperature is a critical operating condition that needs to be highly controlled and monitored since it controls gas composition, tar concentration; reaction rate and ash build up (Asadullah, 2014; WIPO,2013). The operating temperature in the gasifier is recommended to be as high as possible (>800°C) (Bhavanam and Sastry 2011; Basu, 2006). Calibration of the bed temperature will help to ensure that the required temperature is attained. Calibration of the reacting vessel will help to ensure that the optimum conditions required for these process parameters to give a high quality gaseous fuel is attained. For any instrument used for process analysis in the industry, the precision and accuracy of such instrument must be determined so as to give credibility to the response obtained from it especially in a newly constructed and installed instrument. Instrument Calibration is intended to eliminate or reduce bias in the reading over a range for all continuous value (Adeeyinwo et al.,2013). . Determining and establishing the temperature at the reaction zone to be as high as required for the operation will bring an appreciable increase in the product gas / gas yield and heating value. Temperature affects both the amount of tar formed and also the composition of tar since the chemical reactions are temperature dependent. (Bharanam and Sastry, 2011).

One of the key technological challenges limiting the reliability and economics of coal and biomass gasification is the lack of temperature sensors capable of reliably performing in a harsh gasification environment over extended periods of operation (Jia et al., 2013). According to World Intellectual Property Organisational International Bureau (2013), it is an object of invention for an operator to accurately and instantaneously adjust and fine-turn process parameter for the gasification of various materials in a gasifier reactor. Innovative attempts and methods have been described by different researchers on how to determine and validate the temperature in a gasifier especially at the reduction zone. Most of these methods are secondary measurements that are relatively easy to obtain such as temperatures, pressures, and compositions of streams into and out of a gasifier can be used in conjunction with empirical or theoretical models and correlations to estimate inaccessible operating parameters inside the reaction zone.

New methods of obtaining direct measurements of gasification temperature using methods that do not require the direct insertion of a fragile sensing element into the harsh environment is gradually taking over (WIPO, 2013). The most widely used techniques in this category are optical measurements, including combustion specific measurements of temperatures and reaction composition with minimally invasive (require a transparent access port) which is a function of the type of gasifier and the field of research. Jia et al., (2013) described a non-invasive ultrasound approach for temperature measurement in a gasifier reacting vessel.

This work describes and presents an initial validation of a simple method requiring minimal access port of a thermocouple and a monitoring of product stream as we adjust and fine-turn other process parameters in a non - slagging gasifier to calibrate the gasifier and ensure high temperature for an optimum generation of the product gas with high combustible content.

1.1 Chemistry of Gasification

Several reactions occur in a downdraft gasifier. These reactions occur in zones as solid and gases move forward in the same direction (Koukouzas et al, 2008) and at different temperature. Basically, when biomass undergo thermochemical conversion in an oxygen deficient environment, the reaction starts with drying in a drying zone at a temperature of about 150° C – 300° C, producing water vapour. This is followed by the pyrolysis zone where the biomass is pyrolysed at a temperature of about 300° C to 500° C to give char, tar and a mixture of gases Eqn (1). Biomass _____ Char + tar and gases (CO₂, CO, H₂O, H₂, CH₄, CnHn) ------ Eqn (1)

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Combustion reaction of the pyrolysis product occurs in the oxidation (combustion) zone producing secondary reaction and heat needed to carry out the whole process.

 $C + O_2 \longrightarrow CO_2 \quad \Delta H = -408 \text{KJ/mol}$ (2) $2C + O_2 \qquad 2CO \implies H = -246 \text{KJ/mol}$ (3)

These secondary reaction of the pyrolysis and oxidation gases pass through to the reduction zone where actual gasification occurs. Here there is a reduction of the pyrolytic char and a temperature of about 700 $^{\circ}$ C - 1000 $^{\circ}$ C. The end product of the chemical reactions that take place in the reduction zone is a combustible gas which can be used as fuel gas.

This gives room for reduction in tar production (<100mg/Nn) (Bala et al., 2007, Basu, 2013 / 2018). Most of the heat produced is used up at this zone. If the residence time in the hot zone is too short or the temperature too low, then medium sized molecules can escape and will condense as tars and oils, in the low temperature parts of the system.

 $C + CO_2 \longrightarrow 2CO \quad \Delta H = +172 \text{KJ/Mol} \quad (4)$ $C + H_2O \longrightarrow CO + H2 \quad \Delta H = +131 \text{KJ/Mol} \quad (5)$

2. EXPERIMENTAL MATERIALS METHODS

Biomass wood waste (sawdust) collected from sawmills within Akure metropolis, Ondo State, Nigeria were pelletized and used for the calibration of the gasifier. Sawdust was discovered as the most abundant wood waste type in most sawmills within and around Akure which require useful conversion from waste to wealth. These samples were initially dried, sorted into particle sizes using a sieve shaker. To achieve uniform temperature and better heat transfer in the particles lower size of 600 μ m retained was selected for further work (Cao et al., 2008; Malik et al.,2015). The sample were made into pellets having a shape factor of 2.7 (Biswas and Umeki, 2005; Garcia- Maraver and Carpio, 2015). The samples were characterized prior to pelletisation and the basic physiochemical properties that can affect the process was determined. This led to the introduction of binder in the pellet production. The moisture content was determined to meet the required optimum of 10 – 15% as stated by most research to reduce the energy that will be lost in the gasifier during drying. (Aashish and Sumeet, 2016). The result of the physicochemical characterization has been reported (manuscript under review).

2.1 Experimental Set up

A fixed bed downdraft gasifier designed and constructed in – house was calibrated for the biomass gasification process. The schematic diagram is shown in Figure 1. The installation is composed of the following parts ; biomass feeding system, air pump, gasification reactor vessel (drying zone, oxidation, pyrolysis zone, reduction zone, air inlet nozzle called tuyree, gas/solid outlet, ash discharge tray); a flaring unit; cyclone attached to a sample bottle, a free sample bottle; a filter, a pump, outlet port and gas detection set up. (gas cell, water trophy, absorption bottles).Temperature monitoring unit. These parts are arranged in modules for easy of adjustment and adaptation

The gasification vessel is a furnace system having a throated constriction supported in a cylindrical vessel. The reduction zone lagged with fibre glass as the refractive material to ensure high temperature and high reduction of char at this zone. It has one air inlet tuyree opening through the air pump to atmospheric pressure. A grate attached at the bottom of the chamber provides support for the bio- fuel. A type K thermocouple was attached to the reduction zone of the system. The feeding system is attached to the top of the chamber. A discharge tray is attached at the bottom of the chamber to collect solid residue.

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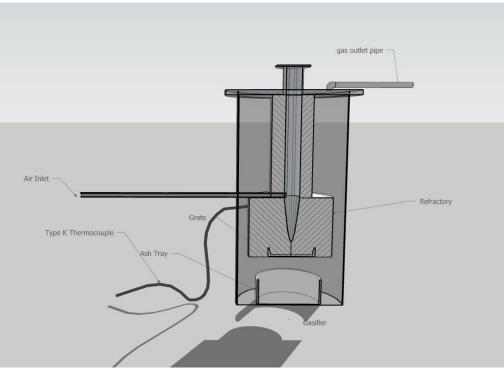


Figure 1. : Gasification reaction vessel

2.3 Experimental Method

The initial operation trials were done with air as gasification agent supplied by an air pump of 130W capacity with a flow rate of approximately $9.16 \times 10^{-6} \text{m}^3/\text{h}$ introduced through a side tuyree which was left at a maximum and the feed inlet was left open at a maximum such that the equivalent ratio ER = 1 (100%). The temperature of the reacting vessel and the process was monitored and calibrated using a type K thermocouple. The thermocouple was calibrated using standard chemicals of known melting point before the various trial tests began (Cao et al., 2008; Igboanugo, 2003). The feedstock was further dried after the initial operations with some system adjustments.

The system adjustments were made with an increase in the fiber glass at the reduction zone for proper refractory; a reduction in the air flow ratio at the inlet and an opening of the feed inlet; an introduction of char from the previous run as a bed material. For each stage, the temperature of the system was monitored and the content of the sample bottles observed and the ash discharge system monitored. These trials were repeated severally to ensure that the operating conditions needed to achieve optimum conversion to combustible gas products is achieved.

3. **RESULTS AND OBSERVATIONS**

The result of the initial calibration of the thermocouple presented in Table 1 showed that the thermocouple is good and ready for use in the gasification process. The result of the temperature calibration process is presented in Table 2- 4. Gasification process is influenced by a number of variables which include particle size, moisture content and chemical composition of the wood type such as ash content (feed stock variables). This was observed from the large water vapour that was present in the sample bottle at the initial stage that led to further drying of the feed sawdust pellets. The process variables like temperature, air ratio, pressure, bed material (Link et al., 2013) was also observed and adjusted in the process. As the inlet air was given 100% complete opening such that equivalent ratio ER = 1, temperature increased and combustion was at a maximum and short resident time was observed but slagging and large ash deposit was observed. This showed that the melting point of ash was also gotten and this is not helpful in the gasification process (Bhavanam and Sastry, 2011).

The sample bottles had large deposits of tar, oil and water vapour (Plate1) after the first few trials. This confirms that conversion and optimum conditions have not been achieved. When there are non-pyrolyzed fuel, hydrocarbons (tar) will be evolved which have a condensation temperature of $200-500^{\circ}$ C when wood is gasified (Fjellerup et al., 2005) as was shown by the thermocouple reading. When the bed material (refractory lining and char from previous run as catalyst) was increased at the char reduction zone, temperature of the reacting vessel was increased and the air tuyree regulated to be about $\frac{1}{3}$ such that the air flow ratio was reduced. The resident time was also regulated and an obvious

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temperature ramping was observed, the tar component of the gas was reduced drastically with no slagging as witnessed by clear sample bottles Plate 2. (Zhon et al., 2015; Caraccio et al., 2014). The water vapour was also reduced. The type k thermocouple confirmed that the temperature of the vessel went as high as 780 $^{\circ}$ C and above.

The thermocouple reading was dragging and was not rising steadily as the temperature got to 600° C and above. A pyrometer is recommended for such high temperature operation and reading (Jia et al., 2013; WIPO, 2013). The relatively clear sample bottles confirm that the prevalent temperature of the gasification reaction vessel is above the condensation temperature of tar. In the work of Fjellerup et al., (2005), downdraft gasifier at this stage nearly elimates these compounds except for a trace of compound with m/z 60 and the tertiary aromatics are now predominant: benzene (78), naphthalene (128), phenanthrene (178) and pyrene (202). Tertiary alkyl aromatics are also present: toluene (92), indene (116); and further some phenol (94). The composition of the tar was not considered at this stage bearing in mind that the aim of the calibration was to ensure temperature optimization. Further work will on the elimination of these compounds using downstream secondary methods to further improve the quality of the gaseous fuel.

Table 1. Calibration of the thermocouple

Standard Chemicals	Melting point (°C)	Type k Thermocouple (°C)	Observations
Environment	28	28	Warm Day
Benzoic Acid	128	128	White powder turned clear
Solder (alloy of Pb & Sn)	430	423	Melted
Zinc granules	500	500	Melted into ashes
Melting Point glass	560-570	560	Bent
Magnesium ribbon	651	648	Deflagrated with a blinding
Impure black lead (iv) oxide	720	700	Turned yellow Lead (ii) oxide
Silver coil	960	780	

Table 2. Calibration of the Reacting Vessel -- 1st Trial Test

Operational Parameters	Thermocouple	Observation
-	Reading	
		Water vapour
Equvalent ratio $ER = 1$		
Particle Size = $x \le 600 \ \mu m$	250	Dark flowing slag
Refractory Material (fiber glass) - low	300	Deposit of tar through the stream
Bed Material (char/ ash) - None	450	Sample bottles completely dark
% moisture content (dry basis) 10.85%	480	
Ash content of feed $-1.93 \pm$	520	
Resident time – Short	600	
Tar deposit - Much	650	Wall temperature increased
Slagging - Yes	600	
Combustion rate – High	650	



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Plate 2: Sample bottles containing large deposit of Tar at temperatures below 500°C



Plate 3: Sample bottles at temperature between 500°C and above.

Table	e 3.	Calibrat	tion of th	ne Reacting	Vessel	2 nd Trial Test
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Operational Parameters	Thermocouple	Observation
	Reading	
Equivalent ratio $ER = 2/3$	28	Warm day
Particle Size = $x \le 600 \ \mu m$	128	Water vapour
Refractory Material (fiber glass) – Increased	350	Cloud of smoke / water vapour
Bed Material (char/ ash) – Low	400	Dark fluid deposits on the bottle
% moisture content (dry basis) – 10.85%	460	Dark tar deposit
Ash content of feed –	500	Water / oil deposit
Resident time – longer	580	Hot flue gas
Combustion rate – reduced	600	Hot flame /

Table 4. Calibration of the Reacting Vessel --- 3rd Trial Test

Operational Parameters	Thermocouple	Observation
	Reading	
Air flow ratio $ER = \frac{1}{3} (0.3)$	29	Clear
Particle size = $x \le 600 \ \mu m$	200	Cloud of fumes
Refractory Material (fiber glass) - Increased	380	Steam/ Water vapour on walls surface
Bed Material (char/ ash) - Increased	500	Clear surface & golden fluid
% moisture content (dry basis) – 10.85%	600	Golden fluid /red hot flame in the chamber
Ash content of feed – Further drying of feed at 120°C	680	Hot gas /very hot flame /thermocouple reading not moving as before
Resident time – longer	680	Glowing flame at flaring unit
Combustion rate – reduced	600	Glowing flame at flaring unit

CONCLUSION

Experimental temperature calibration of the reaction vessel and the monitoring of the other operational parameters such as the air flow ratio, pellet nature and composition, reaction time, residue on sample bottles and ash discharge tray by adjusting and has helped in achieving standard operating conditions for obtaining a tar-free combustible gas in-situ. Large deposit of tar and slag was observed at temperatures below 500°C. At temperatures above 600°C little or no tar

was observed. This implies that pyrolysis of wood char at temperatures above 600 \degree C did not result in further release of tar compounds from the char as was also confirmed by (Fjellerup et al., 2005). This will help to achieve high level of conversion of raw material to gas. A thermocouple is better used for low temperature operations, a pyrometer is recommended for such high temperature operations. This work presents a simple but valid experimental process for the determination and calibration of temperature in any fixed bed gasifier.

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