

# Mechanical Properties of PVA-ZnO and PVA–CeO<sub>2</sub> nanocomposite films

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**Abstract:** Mechanical properties of pure Poly Vinyl Alcohol (PVA), PVA-9wt% ZnO, PVA-2.5wt% CeO<sub>2</sub> and PVA-25wt% CeO<sub>2</sub> nanocomposite films prepared by traditional solution casting technique were studied using Universal Testing Machine (UTM) to elucidate various parameters like tensile strength, Young's modulus, percentage of elongation at the break to understand the mechanical strength of these materials. Nano ZnO and nano CeO<sub>2</sub> used as fillers were prepared by cost effective low temperature solution combustion method. The crystallinity and surface morphology and nature of dispersion of nano fillers in these films were ascertained by X-ray diffraction (XRD) and Scanning electron microscopy (SEM) respectively. The studies reveal an increase in tensile strength and Young's modulus of the material in nano CeO<sub>2</sub> doped composite films compared to the host PVA polymer matrix. Whereas, nano ZnO doped composite films showed a slightly decreased value compared to pure PVA. The maximum tensile strength and highest Young's modulus is found in PVA-25wt% CeO<sub>2</sub> nanocomposite films, indicating higher mechanical strength which is attributed to special morphological and microstructural changes in these films which is evident from the SEM images.

**Keywords:** Mechanical Properties, Tensile strength, modulus of elasticity, nanocomposites, XRD

## 1. INTRODUCTION

Polymer nanocomposites represent a new alternative to conventionally filled polymers. Because of the Nano sizes of the fillers, filler dispersed nanocomposites exhibit markedly improved optical, dielectric, thermal and electrical properties, increased elastic modulus and strength, outstanding barrier properties and decreased flammability properties when compared to the pure polymers or their traditional composites. Among the various polymers, Polyvinyl Alcohol(PVA) is a very versatile synthetic polymer that has large scale technological applications owing to its polar and semi-crystalline nature. Various research groups have worked towards optical, optoelectronic, electrical, thermal, mechanical and many more properties of Poly vinyl alcohol(PVA) doped with Nanofillers such as ceramics, metals, metal oxides, semiconductors and so on and revealed how a new composite material with improved properties can be realized in various technological applications [1-26].

CeO<sub>2</sub> is an insulating metal oxide in bulk form. However, when used as a nano particle it shows an increase in conductivity, UV-Visible absorbance, good luminescence and so on [25,27,28]. On the other hand, ZnO which is semi-conducting as a bulk has shown also an insulating behaviour in the nano regime [29]. The use of these nanoparticles as a filler to form nanocomposite films has been reported to give rise to improved dielectric, optical, thermal and mechanical properties. In our earlier work we have studied the dielectric, optical, thermal and sensing properties [22-26] of nanocomposite films prepared with the above two nanoparticles doped in the polymer PVA, as a function of the concentration of the filler (5wt%, 9wt%, 13wt% and 17wt% of ZnO) and ((2.5 wt%, 5 wt%, 7.5 wt%, 12.5 wt% and 25 wt% of CeO<sub>2</sub>). The studies revealed PVA-9wt% ZnO exhibiting optimum photoluminescence, maximum conductivity, improvised thermal stability and enhanced sensing behaviour among all the ZnO series nanocomposite films. Whereas, among all the CeO<sub>2</sub> series nanocomposite films, PVA-2.5wt% CeO<sub>2</sub> exhibited optimum values in these properties and also PVA- 25wt% CeO<sub>2</sub> showed good thermal stability owing to their special morphological changes forming as nanoribbon or nano bead like structure evident from the SEM and TEM analysis[25]. Henceforth in our present work, we have prepared nanocomposite films of these concentrations (PVA-9wt% ZnO, PVA-2.5wt% CeO<sub>2</sub> and PVA-25wt% CeO<sub>2</sub>) which exhibit optimum enhancement in their optical, electrical, thermal and sensing behaviour to ascertain their elastic modulus and tensile strength to understand their mechanical behaviour.

## 2. EXPERIMENTAL

Synthesis of nanoparticles, pure PVA and nanocomposite films were carried out using Analar grade materials. To synthesize ZnO nanoparticles, a precursor material Zinc nitrate [Zn (NO<sub>3</sub>)<sub>2</sub>, 6H<sub>2</sub>O] was purchased from CDH Ltd., India and a fuel Oxalyl di-hydrazide (ODH) [C<sub>2</sub>H<sub>6</sub>N<sub>4</sub>O<sub>2</sub>] was prepared in the lab according to the instructions given in the book

[30]. Whereas, for the preparation of CeO<sub>2</sub> nanoparticles, a precursor material Cerium nitrate (Ce(NO<sub>3</sub>)<sub>3</sub>.6H<sub>2</sub>O) and a fuel citric acid (C<sub>6</sub>H<sub>8</sub>O<sub>7</sub>) were purchased from CDH Ltd., India. Polymer Poly vinyl alcohol (PVA) with an average molecular weight of 125,000 and 98% hydrolyzed was procured by Aldrich and was used as received. Prepared nanoparticles were uniformly dispersed in polymer matrix by using Cole-Parmer USA, Ultrasonic Processor. Doubly distilled water was used to make all the reactant solutions.

## 2.1 Synthesis

### 2.1.1 Synthesis of ZnO and CeO<sub>2</sub> nanoparticles

ZnO nanoparticles were prepared using a quicker, low-temperature solution combustion method using Zinc Nitrate as precursor and ODH as unique fuel. Whereas CeO<sub>2</sub> nanoparticles were synthesized using cerium nitrate as an oxidizing agent and citric acid as a fuel. The detailed preparation procedures of both nanoparticles are reported elsewhere [22,25].

### 2.1.2 Synthesis of PVA, PVA- ZnO and PVA- CeO<sub>2</sub> nanocomposite films

Pure PVA, PVA-9wt%ZnO, PVA-2.5wt%CeO<sub>2</sub> and PVA-25wt%CeO<sub>2</sub> nanocomposite films were prepared by cost effective solution casting method. The details of the method are reported in our earlier work. The pure and composite films were cut into pieces suitable for measurement. The thickness of the films was measured accurately by an optical method using an Air Wedge and was found to be 180  $\mu$ m /100  $\mu$ m within the experimental error limit of  $\pm$ 1%.

### 2.1.3 Characterization

The X-ray diffraction (XRD) patterns of Pure PVA, PVA-9wt%ZnO, PVA-2.5wt%CeO<sub>2</sub> and PVA-25wt%CeO<sub>2</sub> polymer nanocomposite films were recorded at room temperature using an X-ray powder diffractometer with CuK $\alpha$  radiation ( $\lambda$  = 1.5406 Å) in the 2 $\theta$  (Bragg angles) range ( $10^\circ \leq 2\theta \leq 80^\circ$ ), to obtain the information related to phase purity, crystallinity and details of their structure. The in-depth X-ray diffraction study of pure PVA and both nano ZnO and nano CeO<sub>2</sub> doped composite films were reported elsewhere [22, 25]. Here the analysis repeated for these particular samples reconfirmed the earlier results. The surface roughness and morphology of these samples were analyzed using a scanning electron microscope (ULTRA 55). A micro universal testing machine (LLOYDS—5 KN, London, UK) was used to measure tensile strength, elongation percentage (%), and modulus of elasticity. The tests were carried out according to ASTM D-882 standard test (ASTM, 1992) and calculated using NEXYGEN Plus software. Square shaped samples of films 0.0027m<sup>2</sup> area were taken for the determination of tensile properties.

## 3. RESULTS AND DISCUSSION

### 3.1 X-ray diffraction Analysis

The depth analysis of X-ray diffraction spectra of Pure PVA, PVA-ZnO nanocomposites for various concentrations of nano ZnO and PVA-CeO<sub>2</sub> nanocomposites for various concentrations were reported in our earlier work [22, 25]. Here again, the XRD spectra of samples prepared for the study of mechanical properties were recorded and found to agree with the earlier results. The XRD pattern of pure PVA shown in Figure 3.1(a), exhibits a relatively broad peak centred on 2 $\theta$  = 20.10° with (1 0 1) plane of crystalline PVA and two more peaks at 2 $\theta$  = 27.85° and 2 $\theta$  = 29.20° indicating the semi-crystalline nature of the PVA [22].

Figure 3.1(b) showing the XRD pattern of PVA–9wt% ZnO nanocomposite films indicates many additional peaks when compared with pure PVA, which are attributed to the reflection planes of the ZnO hexagonal Wurtzite crystal structure [22]. These peaks assure the formation of PVA– ZnO hybrid composite with the development of high crystallinity in the PVA matrix due to the complexation of metal oxide nanoparticles with it.

The XRD patterns of PVA-2.5wt%CeO<sub>2</sub> (Fig. 3.1c) and PVA-25wt%CeO<sub>2</sub> (Fig. 3.1d) nanocomposite films show many additional peaks when compared to pure PVA, some of which corresponds to peaks in the XRD of nano CeO<sub>2</sub> [25] establishing the complexation of metal oxide nanoparticles with PVA in the hybrid films.

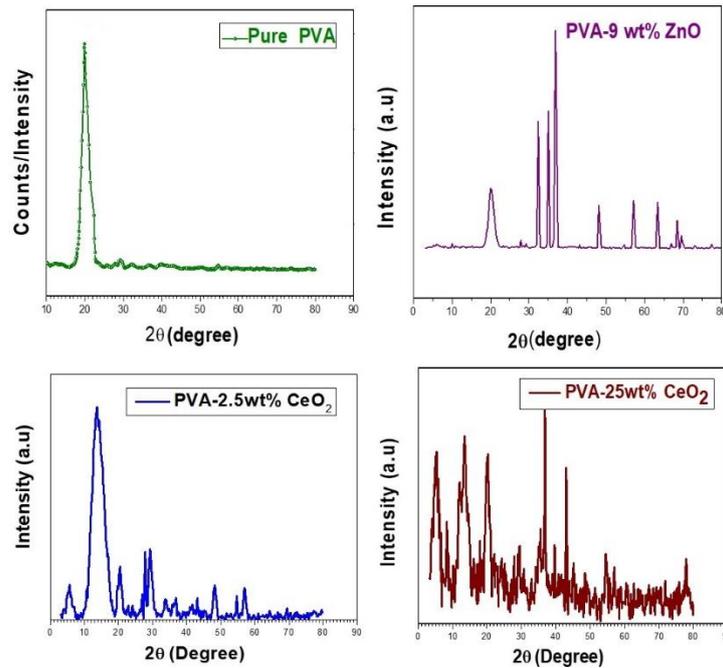


Figure 3.1. XRD spectra of pure PVA and nanocomposite films

Figure 3.1(c) shows an increase in the degree of crystallinity of the nanocomposite with the inclusion of filler  $\text{CeO}_2$ . Whereas, a composite film with the highest concentration of nanofiller (25wt%  $\text{CeO}_2$ ) exhibit a larger decrease in intensity of these lines indicating to be non-existent while the visible peaks correspond to PVA alone. These have also increased in number pointing to the complexation of  $\text{CeO}_2$  with PVA. The results reconfirm the reported analysis elsewhere [22, 25].

### 3.2 Scanning electron microscopy

The SEM micrographs of pure PVA film, PVA– 9wt% ZnO film, are shown in Figure 3.2. SEM of pure PVA at low magnification shows (Figure 3.2(a)) the semi-crystalline nature of PVA supporting the observations of XRD analysis. Whereas hybrid film reveals the uniform distribution of filler nano ZnO in hos polymer PVA matrix confirming the interaction between the two.

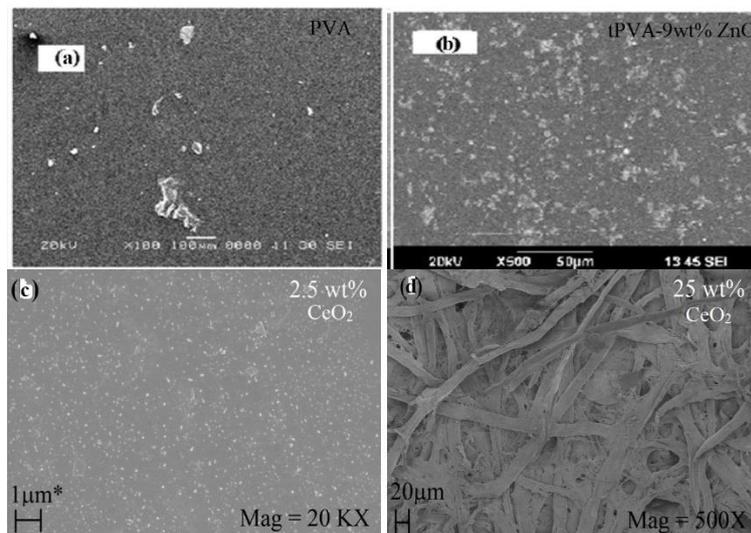


Figure 3.2 SEM spectra of pure PVA and nanocomposite films.

The SEM images of PVA-2.5wt% $\text{CeO}_2$  show uniformly dispersed nano  $\text{CeO}_2$ , with a relatively rough surface compared to pure PVA. The 25wt%  $\text{CeO}_2$  concentration Figure 3.2(d) looks different from the other films with some fibrous tape

like structures. This structure looks like a precursor to nanoribbons along with a porous nature and is seen to affect the mechanical properties investigated in these films. The results observed in SEM spectra of all these samples is similar to detailed analysis carried out in various concentrations of ZnO and CeO<sub>2</sub> doped PVA hybrid films reported elsewhere and also validates the repeatability of results.

### 3.3 Mechanical properties

Mechanical properties were studied to explore the tensile properties of pure PVA, ZnO and CeO<sub>2</sub> doped composite films by using Universal Testing Machine (UTM). The sample films were placed between two metallic grips, wherein the lower grip was held stationary and the upper grip moves upward with the consistent extension of 2.5mm/minute such that constant initial grip separation was maintained for all the samples. The constant speed of the upper grip was recorded by an automatic speed controller fixed to it. The UTM is an electrically driven system. All measurements were performed at ambient temperature conditions in the air medium.

To ascertain the good quality of the material for its long-lasting performance or failure, various tensile parameters like tensile strength, Young's modulus (modulus of elasticity), elongation percentage was measured which links its chemical structure with the exhibited mechanical properties of the samples.

The observation from the Figure. 3.3 and Table 1 reveals the existence of the highest tensile strength in PVA-25wt% CeO<sub>2</sub> nanocomposite films and lower strength in PVA-9wt% ZnO nanocomposite films compared to pure PVA. The highest tensile strength is attributed to a high molecular weight of the polymer which brings in the effect of entanglement. The morphological changes and the extent of increase in crystallinity in these composite films account for the significant effects on the mechanical behaviour of the polymer films. The increase in crystallinity due to secondary bonding, in turn, increases stiffness and elastic modulus and strength of the polymer. The maximum percent total elongation at fracture was found for PVA doped with 2.5wt% CeO<sub>2</sub>. Elongation at break is the highest change in length of a test film before breaking. The results indicate the significant increase in mechanical strength of the host polymer PVA when it is doped with 25wt% CeO<sub>2</sub> evident by an increase in tensile strength, maximum stiffness and Young's modulus as depicted in Figure 3.3 and Table1.

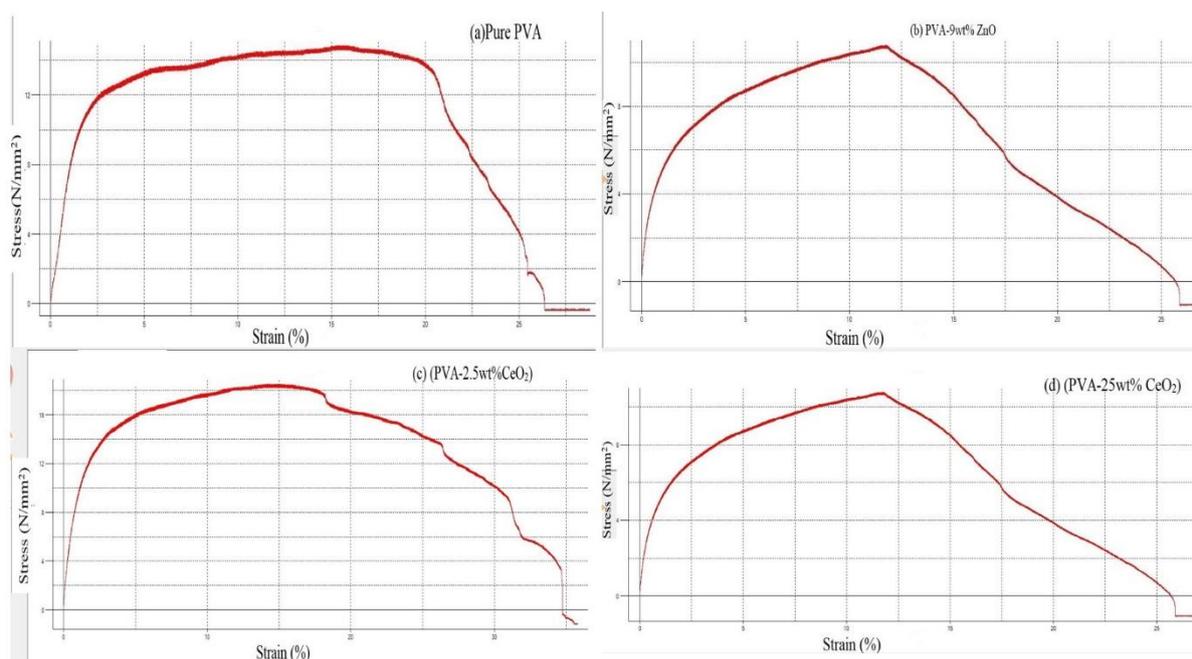


Figure 3.3. Variation of Stress versus strain percent of pure PVA and nanocomposite films.

Table 1. Tensile properties of pure PVA and nanocomposite films:

Sample	Tensile strength (Mpa)	Stiffness (kN/m)	Young's modulus (Mpa)	Percentage (%) total elongation at fracture
Pure PVA	14.84	7.35	259.85	24.41
PVA-9wt% ZnO	10.78	12.46	213.07	14.57
PVA-2.5wt% CeO <sub>2</sub>	18.55	34.78	742.01	23.63
PVA-25wt% CeO <sub>2</sub>	18.00	57.97	1236.49	12.46

ZnO in the nano regime has shown lower mechanical strength as reported by various researchers [31]. Thus inclusion of nano ZnO in PVA has shown decreased tensile strength and lower Young's modulus compared to pure PVA which is revealed from the tabulated values.

### CONCLUSIONS

The nanocomposites films prepared for various concentrations of ZnO and CeO<sub>2</sub> doped to polymer PVA were studied for their morphological and crystallinity properties using SEM and XRD. Among PVA-9wt%ZnO, PVA-2.5wt%CeO<sub>2</sub> and PVA-25wt%CeO<sub>2</sub> polymer nanocomposite films, PVA-25wt%CeO<sub>2</sub> polymer nanocomposite films, exhibit the highest tensile strength and Young's modulus confirming the greater mechanical strength of the nanocomposite films compared to pure PVA.

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### REFERENCES

- [1] Lee J., Bhattacharya D., Eastale A. J., and Metson J. B. "Properties of nano-ZnO/poly(vinyl alcohol)/poly(ethylene oxide) composite thin films." *Curr. Appl. Phys.* **8.1** (2008): 42-47 <https://doi.org/10.1016/j.cap.2007.04.010>.
- [2] Mu Aga, Doping of polymers with ZnO nanostructures for optoelectronic and sensor applications, in: Nicoleta Lupu (Ed.), *Nanowires Science and Technology*, Fisk University, Nashville, TN, 2010, 402.
- [3] Abdullah T., Morimoto M. and Okuyama K. "Generating Blue and Red Luminescence from ZnO/Poly(ethylene glycol) Nanocomposites Prepared Using an In-Situ Method." *Adv. Funct. Mater.* **13-10** (2003): 800. <https://doi.org/10.1002/adfm.200304330>
- [4] Zheng J., Siegel R.W., Toney C.G. "Polymer crystalline structure and morphology changes in nylon-6/ZnO nanocomposites." *J. Polym. Sci., Part B: Polym. Phys.* **41.10** (2003): 1033-1050 <https://doi.org/10.1002/polb.10452>.
- [5] Fernandes D. M., Winkler Hechenleitner A. A., Lima S.M., Andrade L.H.C., Caires A.R.L. and Gomez Pineda E.A., "Preparation, characterisation and photoluminescence study of PVA/ZnO nanocomposite films." *Mater. Chem. Phys.* **128.3**(2011):371-376. [tp://dx.doi.org/10.1016/j.matchemphys.2011.03.002](http://dx.doi.org/10.1016/j.matchemphys.2011.03.002).
- [6] Bouropoulos N., Psarras G.C., Moustakas N., Chrissanthopoulos A. and Baskoutas. S. "Optical and dielectric properties of ZnO-PVA nanocomposites" *Phys. Status Solidi A.* **205.8** (2008): 2033-2037. <http://dx.doi.org/10.1002/pssa.200778863>.
- [7] Hassen A., El Sayed A. M., Morsi W. M., and El-Sayed S. "Influence of Cr<sub>2</sub>O<sub>3</sub> nanoparticles on the physical properties of polyvinyl alcohol." *J. Appl. Phys.* **112.9** (2012): 093525-093525-8. <https://doi.org/10.1063/1.4764864>.
- [8] Seeta Rama Raju G. and Buddhudu S. "Photoluminescence Analysis of Sm<sup>3+</sup> and Dy<sup>3+</sup> Doped PVA Films." *Appl. Polymer Sci.* **107.4** (2008): 2480-2485. DOI 10.1002/app.27257
- [9] Dali Shao, Liqiao Qin and Shayla Sawyer. "Optical properties of polyvinyl alcohol (PVA) coated In<sub>2</sub>O<sub>3</sub> nanoparticles." *Optical Materials.* **35.3** (2013): 563-566. <https://doi.org/10.1016/j.optmat.2012.10.026>
- [10] Chandrakala H. N., Ramaraj B., Shivakumaraiah, and Siddaramaiah. "Optical properties and structural characteristics of zinc oxide-cerium oxide doped polyvinyl alcohol films." *J. of Alloys and Compounds.* **586** (2014): 333-342. <https://doi.org/10.1016/j.jallcom.2013.09.194>.
- [11] Johann Boucle, Punniamorthy Ravirajan and Jenny Nelson. "Hybrid polymer-metal oxide thin films for photovoltaic applications." *J. Mater. Chem.* **17.30** (2007): 3141-3153. 10.1039/b706547g.
- [12] Mohan V. M., Weiling qiu, Jie shen, and Wen chen. "Electrical properties of poly(vinyl alcohol) (PVA) based on LiFePO<sub>4</sub> complex polymer electrolyte films." *J. polym. Res.* **17** (2010): 143-150. DOI 10.1007/s10965-009-9300-0.
- [13] Zhang Q. M., Li H. F., Poh M., Feng X., Cheng Z.Y., Xu H.S. and Huang C. "An all-organic composite actuator material with a high dielectric constant." *Nature.* **419** (2002): 284-287. <https://doi.org/10.1038/nature01021>.
- [14] Azizian Y. and Kalandaugh, "Dielectric properties of CdS-PVA nanocomposites prepared by ultrasound-assisted method." *Optoelectronic and Advanced Materials-Rapid Communications.* **4.11** (2010): 1655-1658.

- [15] Yashar Azizian, Kalandaragh, "Impedance spectroscopy (IS) and thermally stimulated discharged current (TSDC) studies on CdSe-PVA nanocomposites prepared by ultrasound-assisted method." *Optoelectronics and Advanced Materials-Rapid Communications*, **4.2** (2010) 174-179.
- [16] Muradov M. B., Abdinov A. Sh., Hajimamedov R. H. and G. M. Eyivazova. "Dielectric properties of nanocomposites based on copper sulfide nanoparticles and a polymer matrix." *Surface Engg. and Appl. Elec. Chem.* **45.129** (2009): 167-170. <https://doi.org/10.3103/S1068375509020161>.
- [17] Subhojyoti Sinha, Sanat Kumar Chatterjee, Jiten Ghosh and Ajith Kumar Meikap. "Dielectric relaxation and ac conductivity behaviour of polyvinyl alcohol-HgSe quantum dot hybrid films." *J. Phys. D: Appl. Phys.* **47.27** (2014): 275301. <https://doi.org/10.1088/0022-3727/47/27/275301>.
- [18] Mondal S. P., Mullick H., Lavanya T., Dhar A., Ray S. K. and Lahiri S. K. "Optical and dielectric properties of junctionlike CdS nanocomposites embedded in polymer matrix". *J. Appl. Phys.* **102.6** (2007): 064305-7. <https://doi.org/10.1063/1.2784017>.
- [19] Sarma S., Baruah K. and Datta P. "Possible Applications of PVA/PbS nanocomposites." *AIP Conf. Proc.* **1276.1** (2010): 316-321. <https://doi.org/10.1063/1.3504318>.
- [20] M. I. ABd-Elrahman, "Synthesis of Polyvinyl Alcohol-Zinc Oxide Composite by Mechanical Milling." *Nanoscale and Microscale Thermophysical Engineering*, **17.3** (2013): 194-203. <https://doi.org/10.1080/15567265.2013.776152>.
- [21] Jiang L., Shen X.-P., Wu J.-L. and Shen K.-C. "Preparation and characterization of graphene/poly(vinyl alcohol) nanocomposites." *J. Appl. Polym. Sci.* **118.1** (2010):275-279. <https://doi.org/10.1002/app.32278>.
- [22] Hemalatha K. S., Rukmani K., Suriyamurthy N. and Nagabhushana B. M. "Synthesis, characterization and optical properties of hybrid PVA-ZnO nanocomposite: A composition dependent study", *Materials Research Bulletin*, 51 2014:438-446. <https://doi.org/10.1016/j.materresbull.2013.12.055>.
- [23] Hemalatha K. S., Sriprakash G., Ambika Prasad M. V. N., Damle R., and Rukmani K. "Temperature dependent dielectric and conductivity studies of polyvinyl alcohol-ZnO nanocomposite films by impedance spectroscopy" *Journal of Applied Physics*, 118.15. 2015: 154103-13. <https://doi.org/10.1063/1.4933286>.
- [24] Hemalatha K. S, Narasimha Parvathikar and Rukmani K., "Influence of ZnO nanoparticles on thermal behavior of Poly Vinyl Alcoho films", 5.5. 2015: 106-115. <http://www.rspublication.com/ijst/index.html>.
- [25] Hemalatha K. S. and Rukmani K. "Synthesis, characterization and optical properties of polyvinyl alcohol-cerium oxide nanocomposite films." *RSC Adv.* 6.78. 2016: 74354-74366. <https://doi.org/10.1039/C6RA11126B>.
- [26] Hemalatha K. S. and Rukmani K. "Poly vinyl alcohol-CeO<sub>2</sub> nanocomposite films: a promising material for NO<sub>2</sub> sensors at high temperatures." *Mater. Res. Express*, 6.8. 2019: 085008 -7, <https://doi.org/10.1088/2053-1591/ab1ae6>.
- [27] Namrata Shukla, Awalendra K. Thakur. "Enhancement in electrical stability properties of amorphous polymer based nanocomposite electrolyte." *J. of Non crystalline Solids*, **357.22-23**.2011:3689-3701. <https://doi.org/10.1016/j.jnoncrysol.2011.06.036>.
- [28] Zhiyong Fan. and Lu Jia G. "Zinc Oxide Nanostructures: Synthesis and Properties." *J of Nanoscience and Nanotechnology*, **5.10**.2005:1561-1573. <https://doi.org/10.1166/jnn.2005.182>.
- [29] E. Helal, N. R. Demarquette, E. David and M. Fréchette, "Evaluation of dielectric behavior of polyethylene/thermoplastic elastomer blends containing zinc oxide (ZnO) nanoparticles for high voltage insulation," 2016 IEEE Electrical Insulation Conference (EIC), 2016: 592-592, doi: 10.1109/EIC.2016.7548672.
- [30] Patil K.C., Hegde M.S., Tanu Rattan and Aruna S.T. "Chemistry of nanocrystalline oxide materials: combustion synthesis." in: *Properties and Applications*, World Scientific, Singapore, 2008: 332.
- [31] Rithin Kumar N. B., Vincent Crasta., Rajashekar F. Bhajantri, and B.M. Praveen. "Microstructural and Mechanical Studies of PVA Doped with ZnO and WO<sub>3</sub> Composites films." *J. of Polym.* 2014. 2014: 1-7. <http://dx.doi.org/10.1155/2014/846140>.